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1 Optoelectronic properties and interband transition of La-doped BaSnO₃ 2 transparent conducting films determined by variable temperature spectral 3 transmittance

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9 Perovskite-structured Ba_{1-x}La_xSnO₃ ($x=0-0.10$) films have been directly grown on (0001)
10 sapphire substrates by a sol-gel method. Optical properties and bandgap energy of the films have
11 been investigated by transmittance spectra from 10 K to 450 K. It indicates that these films exhibit
12 a high transmission of more than 80% in the visible region. With increasing temperature, there is a
13 significant bandgap shrinkage of about 0.5 eV for lightly La doping ($x \leq 0.04$) films. For heavily La
14 doping concentration ($x \geq 0.06$), the bandgap remains nearly stable with the temperature and La
15 composition. This is due to the fact that the lattice expansion caused by La doping is close to the
16 saturation for the film doped with $x = 0.06$. Moreover, temperature dependent conductivity behav-
17 ior shows a similar pattern, which suggests that the doping concentration of La-doped BaSnO₃
18 (BLSO) films has a saturated state. The La introduction can modify the Sn 5s-O 2p antibonding
19 state and the nonbonding O 2p orbital, which remarkably affect the electronic bandgap of the
20 BLSO films. © 2015 AIP Publishing LLC. [<http://dx.doi.org/10.1063/1.4914482>]

21 I. INTRODUCTION

22 Transparent conducting oxides (TCOs), combining a
23 high optical transparency in visible wavelengths with high
24 electrical conductivity and low conductive resistivity, have a
25 wide range of applications in many fields such as solar
26 energy, flat panel displays, and smart coatings.^{1,2} Many ox-
27 ide materials, such as ZnO, SnO₂, and In₂O₃, have been
28 widely investigated. However, these well-known material
29 systems still have some limitations such as instability of oxy-
30 gen content, fatigue, and degradation. Thus, there has been a
31 challenge to find an appropriate alternative transparent materi-
32 al to exhibit better properties. Fortunately, perovskite-
33 structured oxides represent an important class of transparent
34 materials due to good properties such as ferroelectricity, fer-
35 romagnetism, superconductivity, and piezoelectricity.³⁻⁶ At
36 the same time, multilayer structure of perovskite materials
37 for microelectronic materials and devices also have great
38 potential applications. Therefore, there has been a great
39 necessity for the further study on its structure and the phys-
40 ical mechanism.

41 Alkaline earth stannates with the general formula
42 RSnO₃ (R = Ba, Sr, and Ca) are important transparent mate-
43 rials due to interesting physical properties and perovskite
44 structures.⁷⁻⁹ Here, R is generally rare earth ions or alkaline
45 earth elements. Among RSnO₃, BaSnO₃ (BSO) is an ideally
46 cubic perovskite-type oxide. It behaves as an *n*-type semi-
47 conductor, which has a wide bandgap and exhibits a high
48 transmittance of more than 85% in the visible region.
49 BaSnO₃ has been used in various fields such as physical

50 sensors, epitaxial structure, and future photovoltaic
51 technology.¹⁰⁻¹³ Recent research shows that BSO doped
52 with lanthanum (La) shows great advantages of thin film for-
53 mation. It has been reported that La-doped BaSnO₃ (BLSO)
54 has a high mobility of 320 cm² V⁻¹ s⁻¹ for carrier concentra-
55 tion of 8 × 10¹⁹ cm⁻³ and a bandgap of more than 4 eV,
56 which is significantly larger than those from typical transpar-
57 ent conductive oxides.^{14,15} Consequently, BLSO is a promis-
58 ing candidate for transparent conductor applications and
59 epitaxial multilayer devices. Recently, Shan *et al.* have
60 explored an intrinsic relationship between optical properties
61 and La concentration for BLSO system.^{16,17} There have also
62 been some works on electronic band structure and infrared
63 optical phonons of BLSO bulk material.¹⁸ However, temper-
64 ature dependence of optical properties and bandgap energy
65 of BLSO film has not been studied to date. Therefore, it is
66 desirable for us to conduct a more thorough study on the
67 spectral behavior of high-quality BLSO layer system.

68 In this article, we reported that the transmittance spectra
69 of Ba_{1-x}La_xSnO₃ films with La concentration from $x=0$ to
70 0.10 grown on (0001) sapphire substrates by the sol-gel
71 method. Temperature dependence of electronic transitions
72 and optical properties has been systematically discussed. It
73 was found that La concentration has significant effects on the
74 bandgap energy and dielectric response. Experimental and
75 theoretical results have been provided for a better understand-
76 ing on the physical properties, especially those which can be
77 determined in the electronic orbit and lattice structure.

78 II. EXPERIMENTAL DETAILS

79 The transmittance spectra of the BLSO films with La
80 concentration at $x=0.00, 0.02, 0.04, 0.06, 0.08,$ and 0.10

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81 were recorded by a double beam ultraviolet-infrared spec- 111
 82 trometer (PerkinElmer UV/VIS Lambda 950) at the photon 112
 83 energy of 2650–190 nm (0.5–6.5 eV) with an interval of 113
 84 2 nm. The samples are mounted on a cold stage of an optical 114
 85 cryostat (Janis SHI-4-1) and a heating stage (Bruker A599) 115
 86 for low temperature and high temperature experiments, 116
 87 respectively. Note that the temperature can be varied from 117
 88 10 K to 450 K. For the electrical properties, two-point mea- 118
 89 surement was applied to measure the resistivity of BSLO 119
 90 films. A Keithley 6430 source meter was applied to provide 120
 91 an external direct current with the value of 100 nA to the 121
 92 sample via platinum electrodes. Correspondingly, the resist- 122
 93 ance value can be recorded. 123

94 To extract the dielectric functions and physical param- 124
 95 eters of the BLSO films, the experimental spectra were ana- 125
 96 lyzed by a three-layered structure (air/BLSO/substrate). 126
 97 At near-normal-incidence configuration, the following 127
 98 form describes the transmittance coefficient $t = t_{01}t_{12}e^{-i\delta} /$ 128
 99 $(1 + t_{01}t_{12}e^{-2i\delta})$, where the partial transmittance coefficient 129
 100 t_{01} (vacuum-film) and t_{12} (film-substrate) are written as $t_{i,i+1}$ 130
 101 $= 2\sqrt{\tilde{\epsilon}_i} / (\sqrt{\tilde{\epsilon}_i} + \sqrt{\tilde{\epsilon}_{i+1}})$ and the phase factor for the film with 131
 102 thickness d is described by $\delta = 2\pi d\sqrt{\tilde{\epsilon}_1} / \lambda$. Here, λ is the 132
 103 incident wavelength, and the dielectric functions of vacuum, 133
 104 the film and the substrate are $\tilde{\epsilon}_0 (=1)$, $\tilde{\epsilon}_1$, and $\tilde{\epsilon}_2$, respec- 134
 105 tively. Thus, the spectral transmittance can be readily 135
 106 obtained from $T = \text{Real}(\sqrt{\tilde{\epsilon}_2})tt^*$. It should be noted that the 136
 107 absorption from the substrate must be taken into account to 137
 108 calculate the transmittance of the film-substrate system. For 138
 109 the undoped film, the spectra were fitted with double Tauc-
 110 Lorentz (TL) oscillators. For the doped films, the

transmittance can be modeled with double Tauc-Lorentz 111
 oscillators and a Drude oscillator. The imaginary part of the 112
 Tauc-Lorentz model can be written as $\tilde{\epsilon}_1(E) = \epsilon_\infty$ 113
 $+ \frac{2}{\pi} P \int_0^\infty \frac{\xi \tilde{\epsilon}_2(\xi)}{\xi^2 - E^2} d\xi$, and the real part is given by the Kramers- 114
 Krönig transformation $\tilde{\epsilon}_2(E) = \frac{AE_0C(E-E_g)^2}{(E^2-E_0^2)^2 + C^2E^2} \frac{1}{E} (E > E_g)$, 115
 $\epsilon_2(E) = 0 (E \leq E_g)$. Here, A is the transition matrix element, 116
 E_0 is the peak transition energy, C is the broadening term, E_g 117
 is the bandgap energy, and ϵ_∞ is the high frequency dielec- 118
 tric constant.¹⁹ The Drude model describes free carrier 119
 effects on the dielectric response, whose form is a Lorentz 120
 model with zero center energy. It can be written as $\tilde{\epsilon}(E)$ 121
 $= -\frac{D_n B_r}{E^2 + iB_r E}$. Here, D_n is the oscillator strength and B_r is the 122
 broadening term. 123

III. RESULTS AND DISCUSSION 124

A. NIR-UV transmittance spectra and dielectric 125 functions 126

The insets of Figure 1 show the experimental transmittance 127
 spectra and the best-fitting results of the BLSO films, 128
 which exhibit a high transparency of more than 80% in the 129
 visible region. The best-fitting parameter values of the Tauc- 130
 Lorentz and Drude model with the error bars are summarized 131
 in Table I. In the high-energy region (around 4.0 eV), the 132
 transmission decreases greatly. The spectral loss can be 133
 resulted from the following physical processes: the lattice 134
 expansion and lattice vibrational state. Figure 1 shows the 135
 transmittance spectra of the BLSO films from 10 K to 450 K. 136
 As the temperature increases, the distance between the 137

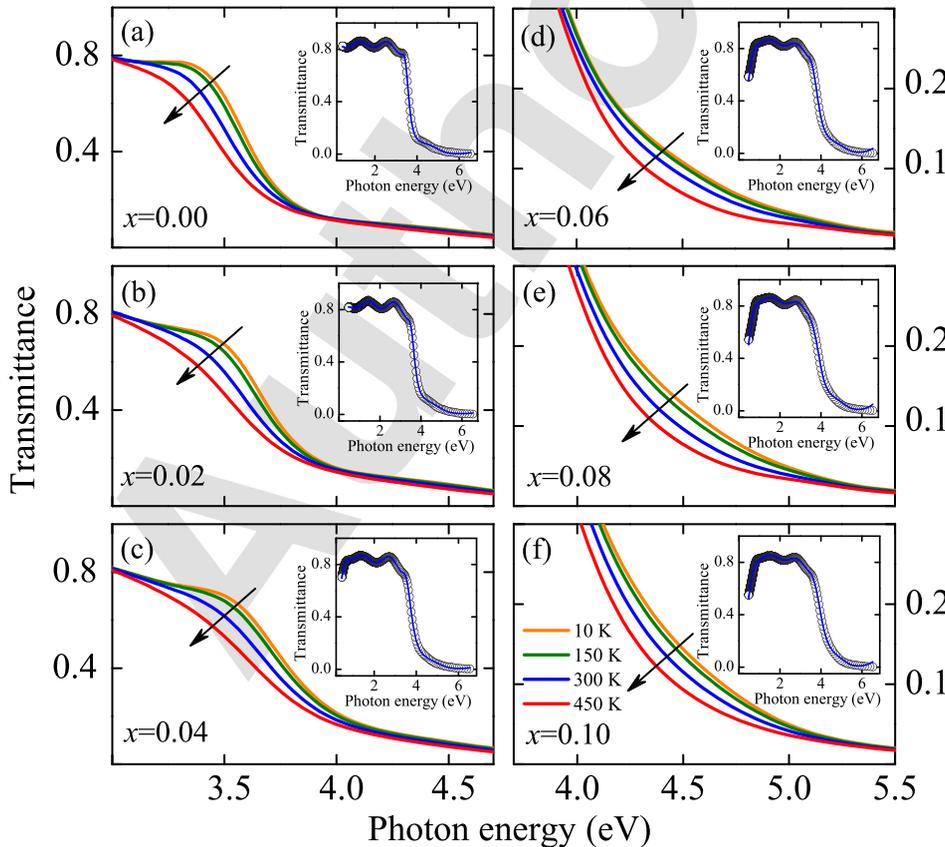


FIG. 1. The transmittance spectra near the absorption region from the BLSO films with different La doping concentration. The arrows show that the absorption edges of the BLSO films present a red-shift trend with the temperature. Note that experimental transmittance spectra (dotted lines) and the best-fitting results (solid lines) of the BLSO films were shown in the insets.

TABLE I. Parameters of the Tauc-Lorentz and Drude dielectric function model for $\text{Ba}_{1-x}\text{La}_x\text{SnO}_3$ films determined from the simulation of the transmittance spectra. The 90% reliability is given in parentheses.

Samples	$x = 0.00$			$x = 0.02$			$x = 0.04$			$x = 0.06$			$x = 0.08$			$x = 0.10$		
	T (K)	10	300	450	10	300	450	10	300	450	10	300	450	10	300	450	10	300
ϵ_∞	2.38 (0.06)	3.15 (0.01)	3.03 (0.02)	3.41 (0.02)	3.18 (0.01)	3.10 (0.02)	3.22 (0.02)	3.12 (0.01)	3.00 (0.03)	2.97 (0.02)	2.72 (0.11)	2.67 (0.21)	2.62 (0.03)	2.54 (0.03)	2.21 (0.01)	2.70 (0.03)	2.41 (0.05)	2.29 (0.02)
A_1 (eV)	28.3 (0.55)	34.1 (0.45)	51.5 (0.61)	17.6 (0.43)	23.3 (0.24)	36.5 (0.71)	19.1 (0.59)	24.5 (0.60)	37.5 (0.63)	12.8 (0.82)	16.3 (0.18)	17.8 (0.21)	11.1 (0.14)	12.3 (0.21)	14.2 (0.38)	11.6 (0.54)	15.1 (0.84)	18.2 (0.32)
E_{01} (eV)	3.77 (0.01)	3.56 (0.01)	3.32 (0.02)	3.86 (0.02)	3.68 (0.01)	3.42 (0.02)	3.95 (0.01)	3.74 (0.01)	3.41 (0.05)	4.17 (0.02)	4.10 (0.02)	4.01 (0.02)	4.20 (0.01)	4.13 (0.01)	4.04 (0.01)	4.22 (0.02)	4.14 (0.02)	4.06 (0.01)
C_1 (eV)	0.99 (0.02)	0.97 (0.01)	0.86 (0.01)	0.84 (0.02)	0.96 (0.01)	1.01 (0.02)	1.20 (0.03)	1.24 (0.02)	1.26 (0.05)	1.38 (0.10)	1.60 (0.12)	1.56 (0.11)	1.42 (0.15)	1.47 (0.15)	1.48 (0.15)	1.39 (0.18)	1.44 (0.17)	1.45 (0.17)
E_{g1} (eV)	3.24 (0.01)	3.24 (0.01)	3.24 (0.01)	3.26 (0.01)	3.26 (0.01)	3.26 (0.01)	3.30 (0.01)	3.30 (0.01)	3.30 (0.01)	3.21 (0.04)	3.21 (0.04)	3.21 (0.04)	3.19 (0.07)	3.19 (0.07)	3.19 (0.07)	3.39 (0.06)	3.39 (0.06)	3.39 (0.06)
A_2 (eV)	17.9 (0.94)	7.35 (0.15)	9.10 (0.39)	7.85 (0.23)	10.2 (0.12)	12.0 (0.43)	6.00 (0.24)	7.98 (0.19)	9.72 (0.47)	1.89 (0.15)	1.79 (0.15)	1.59 (0.15)	1.96 (0.19)	1.83 (0.19)	1.62 (0.19)	3.18 (0.26)	2.90 (0.24)	2.72 (0.25)
E_{02} (eV)	5.78 (0.03)	5.82 (0.04)	5.76 (0.11)	5.65 (0.03)	5.79 (0.02)	5.80 (0.09)	5.80 (0.04)	5.83 (0.04)	5.78 (0.12)	5.84 (0.06)	5.81 (0.07)	5.83 (0.08)	5.81 (0.08)	5.81 (0.08)	5.81 (0.09)	5.77 (0.07)	5.71 (0.07)	5.66 (0.07)
C_2 (eV)	0.47 (0.03)	2.70 (0.04)	3.74 (0.19)	1.97 (0.06)	3.18 (0.03)	4.11 (0.18)	1.88 (0.06)	2.79 (0.06)	3.72 (0.25)	1.32 (0.12)	1.46 (0.14)	1.51 (0.16)	1.49 (0.17)	1.53 (0.18)	1.55 (0.20)	1.69 (0.15)	1.72 (0.16)	1.79 (0.19)
E_{g2} (eV)	2.23 (0.01)	2.23 (0.01)	2.23 (0.01)	2.37 (0.01)	2.37 (0.01)	2.37 (0.01)	2.38 (0.02)	2.38 (0.02)	2.38 (0.02)	1.05 (0.26)	1.05 (0.26)	1.05 (0.26)	0.97 (0.41)	0.97 (0.41)	0.97 (0.41)	1.62 (0.19)	1.62 (0.19)	1.62 (0.19)
D_n (eV)	2.41 (0.16)	2.13 (0.05)	2.61 (0.12)	8.57 (0.20)	7.26 (0.07)	8.65 (0.18)	7.09 (0.53)	6.82 (0.49)	7.37 (0.53)	7.28 (0.76)	6.56 (0.69)	7.60 (0.79)	5.07 (0.48)	5.34 (0.44)	5.83 (0.47)
B_r (eV)	0.08 (0.01)	0.09 (0.01)	0.08 (0.01)	0.08 (0.01)	0.09 (0.01)	0.07 (0.01)	0.11 (0.01)	0.12 (0.01)	0.11 (0.01)	0.12 (0.01)	0.13 (0.01)	0.11 (0.01)	0.16 (0.01)	0.15 (0.01)	0.14 (0.01)

138 lattices gradually increases and the lattice wave frequency
 139 changes with the crystal volume. The phenomenon is due to
 140 the fact that there has been an upward movement of the va-
 141 lence band bottom as the electrons of the semiconductor gain
 142 more energy. Another factor is the variation of the state of
 143 lattice vibrations with the temperature, which results in the
 144 variation of the phonon excitation state. Thus, it leads to the
 145 electron-phonon coupling and the variation in the perturba-
 146 tion of energy band. Both of these effects cause a relative
 147 movement of the bandgap edge. Note that these mechanisms
 148 become increasingly important for the photon energy near
 149 the absorption region. The absorption edge presents a typical
 150 red-shift trend with the temperature, which can be observed
 151 from most of semiconductors and dielectrics. For the specific
 152 variation of the absorption edge, there has been a difference
 153 between lightly La doping ($x \leq 0.04$) and heavily La doping
 154 ($x \geq 0.06$). For the La concentration at $x=0.00$, 0.02, and
 155 0.04, temperature dependent variations in the absorption
 156 edge position are around 3.5 eV. For the La concentration at
 157 $x=0.06$, 0.08, and 0.10, however, the variations are located
 158 at about 4.3 eV.

159 The evaluated dielectric function $\varepsilon = \varepsilon_1 + i\varepsilon_2$ of the
 160 BLSO films are shown in Figure 2. The maximum position
 161 of dielectric function has a red-shift trend with increasing
 162 temperature. The real part ε_1 in the low photon energy
 163 increases with the temperature, while it shows an opposite
 164 trend at the high photon energy side. The imaginary part ε_2
 165 shows the similar trend as the real part. Note that both of the
 166 critical point energies, in general, show a red shift with

167 increasing temperature, which can be attributed to the ther-
 168 mal expansion of the lattice and renormalization of the band
 169 structure by electron-phonon interaction.²⁰ From the imagi-
 170 nary part ε_2 , there have been two obvious electronic transi-
 171 tions in BLSO materials. The first ascent of the absorption
 172 edge is around 3.5 eV and the second is around 4.3 eV, which
 173 indicates that there has been another strong absorption. In
 174 terms of the electronic band structure, the conduction band
 175 minimum (CBM) of BaSnO₃ originates from the Sn 5s-O 2p
 176 antibonding state. The valence band maximum (VBM) is
 177 derived from the nonbonding O 2p orbital.²¹ Note that the
 178 CBM and VBM are situated in the Γ point and the R point,
 179 respectively. The first optical absorption (E_{01}) is due to the
 180 direct transition from VBM to CBM at Γ point.²² The second
 181 absorption (E_{02}) edge implies another energy minimum in
 182 the conduction band, which is located in the R point. The
 183 electrons can be excited from the valence band to the con-
 184 duction band at R point if the energy is large enough.

185 From Figs. 2(a) and 2(b), we can note that the parameter
 186 ε_1 increases and ε_2 decreases sharply with increasing photon
 187 energy in the near-infrared region for La doped with $x=0.06$.
 188 However, the parameters remain stable in the same region for
 189 the undoped film. This is because of the free carrier absorp-
 190 tion. With increasing the La concentration, more free elec-
 191 trons have been provided inside the films. When the photon
 192 energy is not large enough, it will not cause the band transi-
 193 tion or exciton absorption. Thus, the transition and absorption
 194 occur in the same energy band. From Figs. 2(c)–2(h), we
 195 can see that the trend of the dielectric functions with the

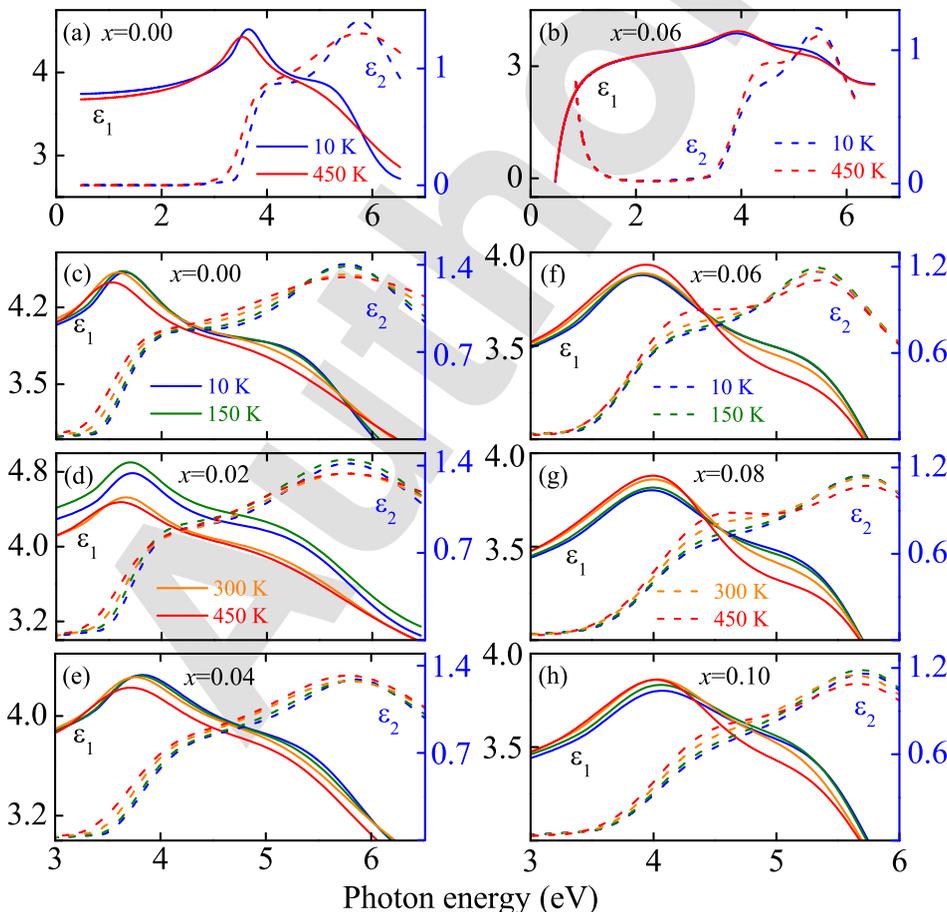


FIG. 2. The real part ε_1 (solid lines) and imaginary part ε_2 (dashed lines) of the dielectric functions for the BLSO films from 10 K to 450 K. Note that the photon energy region is covered from 0.5 to 6.5 eV in (a) and (b), which can present the wide optical dispersion of the BLSO films doped with $x=0.0$ and 0.06.

196 temperature is different from the film doped with $x \leq 0.04$
 197 and $x \geq 0.06$. The real and imaginary parts of dielectric func-
 198 tion increase with the photon energy and approach the maxi-
 199 mum, then decrease upon the photon energy increasing. The
 200 critical point is associated with interband electronic transition
 201 between the valence and conduction bands. For the film
 202 doped with $x \leq 0.04$, two transition regions can be affected
 203 by temperature. However, the second transition can be
 204 affected only for the film doped with $x \geq 0.06$. Thus, the La
 205 concentration of the films affects the energy level of the
 206 materials. At lightly doping ($x \leq 0.04$) film, some La ions can
 207 substitute for Ba ions and the impurity energy level appears
 208 between VBM and CBM. At higher La doping ($x \geq 0.06$)
 209 case, the concentration is close to the saturation and the struc-
 210 ture basically reaches a stable state. Thus, the excess ions
 211 cause little impact on the direct transition.

212 B. Temperature dependent bandgap energy

213 Figure 3(a) shows temperature dependence of the
 214 bandgap energy for the BLSO films. There has been a signifi-
 215 cant difference of the trends between $x \leq 0.04$ and $x \geq 0.06$.
 216 For the films doped with $x \leq 0.04$, the bandgap shows a sharp
 217 decline, corresponding to the temperature increasing from
 218 10 K to 450 K. In particular, it can be found that the bandgap
 219 value decreases from 3.95 to 3.41 eV for the film with
 220 $x = 0.04$. As can be seen, the temperature has an important
 221 role in the variation of the bandgap energy. We can also see
 222 that the rate for variation of the bandgap increases with the
 223 temperature. With increasing temperature, the longitudinal
 224 phonons change the interatomic distance along the direction
 225 of their propagation, and the transversal phonons will be per-
 226 pendicular to their propagation.²³ Thus, the lattice constant
 227 of the BLSO films is varied due to the dilatation of lattice
 228 and the narrowing of interatomic distances. The variation of
 229 the lattice constant results in the modification of the elec-
 230 tronic band structure which further moves the conduction
 231 band downward and the valence band upward.²⁴

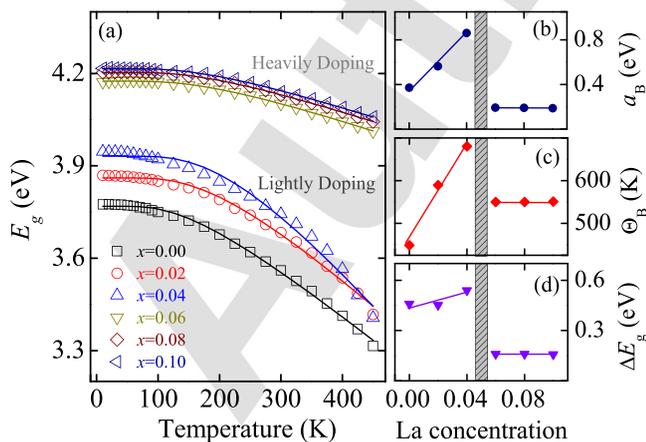


FIG. 3. (a) Temperature dependence of bandgap energy for the BLSO films with different La concentration. The solid lines correspond to the fitting from Bose-Einstein model. The best-fitting parameters (b) a_B and (c) Θ_B of the Bose-Einstein model as a function of doping concentration. (d) The bandgap discrepancy is varied with the La concentration.

232 Generally, temperature dependence of the bandgap param-
 233 eter can be described with the Bose-Einstein model, in which
 234 the carrier-phonon coupling is taken into account.^{25,26} The
 235 model can be written as $E_g(T) = E_g(0) - 2a_B / [\exp(\Theta_B/T) - 1]$,
 236 here, a_B is the strength of the electron-phonon interaction, and
 237 Θ_B is the characteristic temperature representing the effective
 238 phonon energy on the temperature scale.²⁷ The best-fitting pa-
 239 rameters are shown in Figs. 3(b) and 3(c). For lightly La dop-
 240 ing, the parameters a_B and Θ_B increase linearly with La
 241 concentration. For heavily La doping, the parameters do not
 242 change with La concentration. For lightly La doping, the
 243 bandgap variation is more than 0.5 eV, which is much larger
 244 than that (about 0.16 eV) for heavily La doping, as shown in
 245 Figure 3(d). On the other hand, the bandgap increases with
 246 increasing La doping concentration. For different La concen-
 247 tration, the impact factor of the temperature and La doping
 248 concentration on the bandgap energy variation is significantly
 249 different, as shown in Figure 4. For La doping concentration at
 250 $x \leq 0.04$, there has been an obvious variation. As the La doping
 251 concentration increases, there will be more electrons to be
 252 coupled with the phonons, which cause the parameter a_B
 253 decreasing. According to the Bose-Einstein model, the
 254 bandgap energy has a negative correlation with the parameter
 255 a_B , which is consistent with the experimental results. At higher
 256 La doping ($x \geq 0.06$), there is a significant increase of bandgap
 257 energy, which can be explained by the Burstein-Moss effect
 258 due to the degenerate nature of the films.^{28,29} As the carrier
 259 concentration exceeds the Mott critical density of BLSO, the
 260 bottom of the conduction band will be filled. Thus, more
 261 energy is needed to excite the electrons from the donor impu-
 262 rity band to the conduction band, resulting in an optical
 263 bandgap widening.³⁰ For $x = 0.06, 0.08$, and 0.10 , the bandgap
 264 energy still increases with the temperature, but the variation is
 265 much less than that for the La doping concentration at
 266 $x \leq 0.04$. The bandgap energy is relatively stable with temper-
 267 ature. This shows that at a highly doping concentration, the
 268 electronic band and optical properties almost unchanges with
 269 the temperature or La concentration, which can be seen from
 270 the parameters a_B and Θ_B .

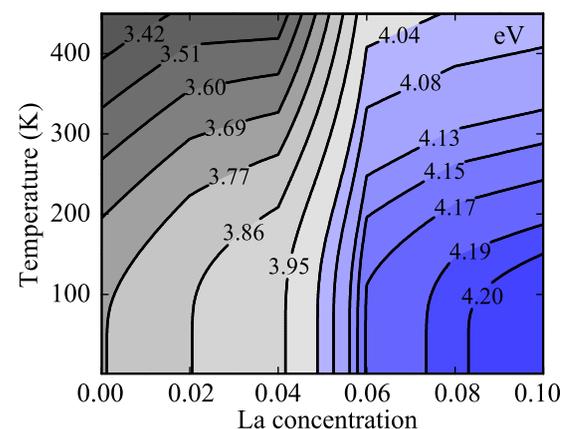


FIG. 4. A two-dimensional picture based on the bandgap energy variation with the temperature and La concentration. The numbers correspond to the bandgap energy and the unit is “eV”.

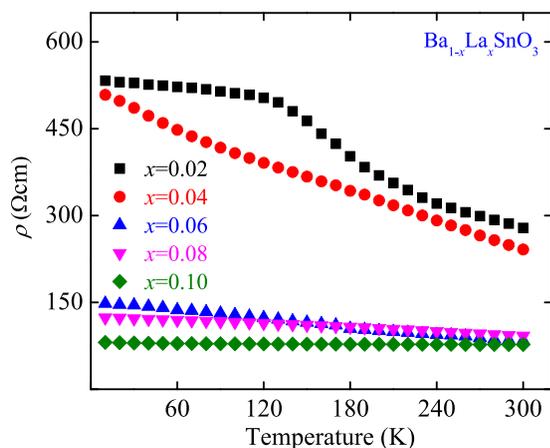


FIG. 5. Temperature dependent resistivity for the BLSO ($x=0.02-0.10$) films. Note that the film doped with $x=0.00$ is not measured because the undoped BSO film is insulating.

271 **C. Electrical resistivity analysis**

272 To further understand the intrinsic characteristics of the
 273 BLSO films, temperature dependence of electrical resistivity
 274 is shown in Figure 5. Note that the undoped film is not measured
 275 because it is insulating. The resistivity decreases systematically
 276 with increasing the temperature from $x=0.02$ to
 277 0.04. It is due to the fact that there have been more free carriers
 278 excited from the donor impurity band to the conduction band
 279 by the thermal excitation. However, it becomes almost
 280 constant for $x \geq 0.06$, indicating that there is a weak temperature
 281 dependence for heavily La doping. The resistivity in
 282 BLSO films is dominated by scattering such as grain boundary
 283 scattering and ionized-dopant scattering. As the carrier
 284 concentration exceeds the Mott critical density of BLSO

285 films, the Fermi level moves up into the conduction band,
 286 which results in a metallic behavior.³¹ It shows that the
 287 energy levels created by substitution of lanthanum for barium
 288 are not significantly stabilized with respect to the bottom
 289 of the conduction band. The variation of the resistivity with
 290 the temperature and La concentration shows a similar pattern
 291 with the bandgap variation, which suggests that the doping
 292 concentration of BLSO films tends to a saturated state.

293 **D. Electronic transition origin**

294 The bandgap variation with La concentration can be
 295 explained by the lattice expansion. As we know, the bandgap
 296 value of a semiconductor is dependent on the position and
 297 width of the conduction band and valence band. The
 298 undoped BSO has an ideal cubic perovskite structure with
 299 the Sn-O-Sn bond angle of 180° . The top of the valence band
 300 depends on the contributions of the O $2p$ orbitals and the bot-
 301 tom of the conduction band is predominantly composed of
 302 nonbonding Sn $5s$ orbitals.³² The ionic radius of La^{3+} ion
 303 is around 0.103 nm, which is less than that of Ba^{2+}
 304 (0.135 nm).³³ Generally, the lattice spacing decreases with
 305 increasing the content of a smaller ion. However, there has
 306 been an unusual lattice expansion for La introduction, as
 307 shown in Fig. 6. With La^{3+} ion doped in BSO, it provides
 308 electron states inside the conduction band of BSO, which is
 309 mainly characterized as the Sn $5s$ states with a Sn-O anti-
 310 bonding character. It means that a La ion acts as an electron
 311 donor and its ionic valence becomes La^{3+} . The occupation
 312 in the antibonding state can cause repulsive force between
 313 Sn and O. It reduces the total energy of the crystal structure,
 314 which results in the expansion of the lattice.¹⁴ The Sn-O
 315 bond length caused by the lattice expansion has been
 316 stretched. The electrons need to gain more energy to transit

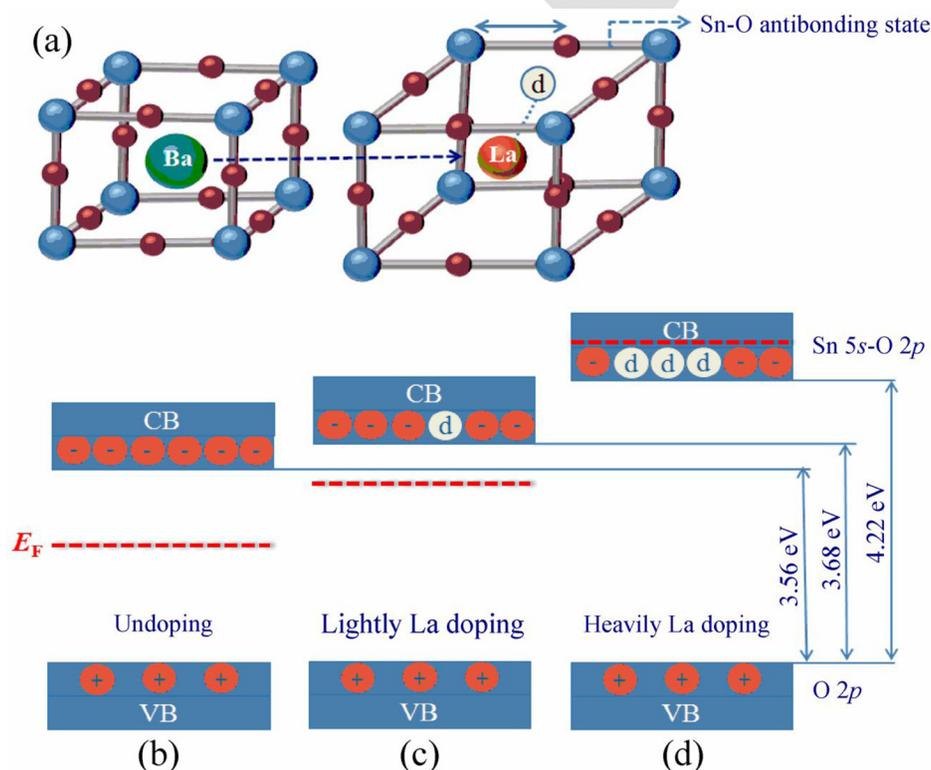


FIG. 6. (a) The cubic structure of the BSO material and the occupation in the Sn-O antibonding state caused by La ion. Schematic representation of the electronic band structure for the BLSO films with the La concentration of (b) 0%, (c) 2%, and (d) 10%, respectively. The parameter E_F denotes the Fermi energy level. Note that the symbols (+) and (-) indicate the intrinsic holes and electrons. The symbol “Ⓢ” indicates the electrons induced by the La-doping.

317 from O $2p$ to Sn $5s$, which explains the increase of the
 318 bandgap energy. The La d orbital energy is high, and the
 319 CBM of BSO has little contribution from Ba ions. Thus, the
 320 substitution on the Ba causes little impact on the CBM.²²
 321 Figs. 6(b)–6(d) show schematic representation for the elec-
 322 tronic band structure of the BLSO films with La concentra-
 323 tion of 0%, 2%, and 10%. For the undoped BSO, the Fermi
 324 level is closed to the bandgap midline. The increment of La
 325 concentration would induce the increase of electron carrier,
 326 which results in the form of the tail state in the bottom of
 327 conduction band. The Fermi level moves up with the
 328 increase of La concentration, and the bandgap energy shows
 329 a blue-shift. For heavily La doping, it induced the Fermi
 330 energy level going into the conduction band, and formed the
 331 Burstein-Moss effect. Each substitution for Ba ion can
 332 donate an electron inside the conduction band of the BLSO.
 333 Also, the Fermi level was formed in the conduction band as
 334 the doped La atom can donate an electron carrier to the con-
 335 duction band by the ionization. It leads to the degenerate
 336 semiconducting behavior of the material.

337 For higher La doping ($x \geq 0.06$), the Fermi level moves
 338 up into the conduction band, thus causing the electron hop-
 339 ping hardly from the minimal energy level to the VBM. It
 340 results in a significant increase in the bandgap. The minimal
 341 band energy is expanded obviously with the La doped with
 342 $x \geq 0.06$. As mentioned before, there are two transitions at Γ
 343 point and R point for the BLSO films. As for heavily La dop-
 344 ing, the Fermi level is above the CBM at Γ point. Thus, the
 345 first absorption E_{01} has a very little probability of occurrence
 346 and the electrons can only be excited from the valence band
 347 to the conduction band at the R point. The bandgap energy is
 348 around 4.0–4.3 eV for heavily La doping, which is consistent
 349 with the transition energy at the R point. Meanwhile, the
 350 doping concentration may tend to a saturated state. The
 351 excess La^{3+} has little chance to replace Ba^{2+} . Thus, the trend
 352 of lattice expansion has a large decrease. The temperature
 353 and La concentration cause little impact on the lattice expan-
 354 sion and/or bandgap energy variation. Recently, high quality
 355 $\text{CuGa}_{0.8}\text{Cr}_{0.2}\text{O}_2$ films (a kind of p -type TCOs) has been
 356 grown by chemical solution technique.³⁴ Its Fermi level is
 357 above the top of the valence band. A prototype of pn junction
 358 can be composed by n -type BLSO and p -type
 359 $\text{CuGa}_{0.8}\text{Cr}_{0.2}\text{O}_2$. The pn junction shows a great potential to
 360 realize the photoelectrochemical cell.
 361 used in fully transparent photoelectrochemical cell.

362 IV. SUMMARY

363 In summary, we report transparent and conductive La-
 364 doped BSO films with La concentration at $x = 0\text{--}0.10$ depos-
 365 ited on (0001) sapphire substrates by a sol-gel method. The
 366 films exhibit a high transmittance in the visible region. To fur-
 367 ther investigate the optical properties of BLSO films, the dielectric
 368 functions and bandgap energy of BLSO films, the dielectric functions
 369 were uniquely extracted by fitting the transmittance spectra
 370 in the photon energy of 0.5–6.5 eV from 10 K to 450 K. It
 371 was found that the absorption edge of the BLSO films
 372 presents a red-shift trend with the temperature. The dielectric
 373 function shows that there have been two obvious electronic

transitions in BLSO (around 3.5 eV and 4.5 eV). It is interest-
 ing to be found that the effect of temperature on the variation
 of bandgap energy and transmittance spectra is different
 between lightly doping ($x \leq 0.04$) and heavily doping
 ($x \geq 0.06$). Because of an unusual expansion, there has been
 a limit of the solid solution for the La^{3+} ion in BSO, which
 is not common in ion doped materials. Thus, there has been
 a saturated doping concentration for this material system.
 For heavily doping, BLSO behaves as a degenerate semicon-
 ductor, which has an obvious variation with the physical
 properties of lightly doping BLSO. The variation of the resis-
 tivity shows a similar pattern with the bandgap, which is
 consistent with the results.

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- ¹G. Thomas, *Nature* **389**, 907 (1997).
- ²K. Hayashi, S. Matsuiishi, T. Kamiya, M. Hirano, and H. Hosono, *Nature* **419**, 462 (2002).
- ³C. H. Ahn, J. M. Triscone, and J. Mannhart, *Nature* **424**, 1015 (2003).
- ⁴O. Auciello, J. F. Scott, and R. Ramesh, *Phys. Today* **51**(9), 22 (1998).
- ⁵B. N. Mbenkum, N. Ashkenov, M. Schubert, M. Lorenz, H. Hochmuth, D. Michel, M. Grundmann, and G. Wagner, *Appl. Phys. Lett.* **86**, 091904 (2005).
- ⁶K. Dörr, *J. Phys. D: Appl. Phys.* **39**, R125 (2006).
- ⁷L. Geske, V. Lorenz, T. Muller, L. Jager, H. Beige, H. P. Abicht, and V. Mueller, *J. Eur. Ceram. Soc.* **25**, 2537 (2005).
- ⁸U. Lampe, J. Gerblinger, and H. Meixner, *Sens. Actuators, B* **26**, 97 (1995).
- ⁹S. W. Tao, F. Gao, X. Q. Liu, and O. T. Sørensen, *Sens. Actuators, B* **71**, 223 (2000).
- ¹⁰U. Lampe, J. Gerblinger, and H. Meixner, *Sens. Actuators, B* **18**, 132 (1994).
- ¹¹S. V. Manorama, C. V. G. Reddy, and V. J. Rao, *Appl. Surf. Sci.* **174**, 93 (2001).
- ¹²Y. F. Zhang, H. R. Zhang, Y. F. Wang, and W. F. Zhang, *J. Phys. Chem. C* **112**, 8553 (2008).
- ¹³Y. Z. Wang, E. Bevilion, A. Chesnaud, G. Geneste, and G. Dezanneau, *J. Phys. Chem. C* **113**, 20486 (2009).
- ¹⁴H. J. Kim, U. Kim, T. H. Kim, J. Kim, H. M. Kim, B. Jeon, W. Lee, H. S. Mun, K. T. Hong, J. Yu, *J. Chem. Phys.* **136**, 114701 (2012).
- ¹⁵X. Luo, Y. S. Oh, A. Sireniko, T. Gao, T. A. Johnson, K. Chai, and S.-W. Cheong, *Appl. Phys. Lett.* **100**, 172112 (2012).
- ¹⁶C. Shan, T. Huang, J. Z. Zhang, M. J. Han, Y. W. Li, Z. G. Hu, and J. H. Chu, *J. Phys. Chem. C* **10**, 1021 (2013).
- ¹⁷C. Shan, P. Chang, K. Shi, Y. W. Li, Z. G. Hu, and J. H. Chu, *RSC Adv.* **04**, 4987 (2014).
- ¹⁸T. N. Stanislavchev, A. P. Litvinchuk, X. Luo, and S.-W. Cheong, *J. Appl. Phys.* **118**, 6994 (2014).
- ¹⁹G. E. Jellison and F. A. Modine, *Appl. Phys. Lett.* **69**, 371 (1996).
- ²⁰H. Y. Fan, *Phys. Rev.* **82**, 900 (1951).
- ²¹S. Sallis, D. O. Scanlon, S. C. Chae, N. F. Quackenbush, D. A. Fischer, J. C. Woicik, J.-H. Guo, S. W. Cheong, and L. F. J. Piper, *Appl. Phys. Lett.* **103**, 042105 (2013).
- ²²H. R. Liu, J. H. Yang, H. J. Xiang, X. G. Gong, and S. H. Wei, *Appl. Phys. Lett.* **102**, 112109 (2013).
- ²³S. Biernacki, U. Scherz, and B. K. Meyer, *Phys. Rev. B* **49**, 4501 (1994).

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[dx.doi.org/10.1021/jp500100a](https://doi.org/10.1021/jp500100a)

The doi is
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118, 6994 (2014)

34987

- 441 ²⁴B. S. Li, Y. C. Liu, Z. Z. Zhi, D. Z. Shen, Y. M. Liu, J. Y. Zhang, and X. 450
 442 W. Fan, *J. Cryst. Growth* **240**, 479 (2002). 451
 443 ²⁵M. J. Han, K. Jiang, J. Z. Zhang, Y. W. Li, Z. G. Hu, and J. H. Chu, *Appl.* 452
 444 *Phys. Lett.* **99**, 131104 (2011). 453
 445 ²⁶W. W. Li, K. Jiang, J. Z. Zhang, X. G. Chen, Z. G. Hu, S. Y. Chen, L. 454
 446 Sun, and J. H. Chu, *Phys. Chem. Chem. Phys.* **14**, 9936 (2012). 455
 447 ²⁷M. Beaudoin, A. J. G. DeVries, S. R. Johnson, H. Laman, and T. Tiedje, 456
 448 *Appl. Phys. Lett.* **70**, 3540 (1997). 457
 449 ²⁸E. Burstein, *Phys. Rev.* **93**, 632 (1954). 458
 459
- ²⁹T. S. Moss, *Proc. Phys. Soc. London, Sect. B* **67**, 775 (1954). 450
³⁰Q. Z. Liu, J. J. Liu, B. Li, H. Li, G. P. Zhu, K. Dai, Z. L. Liu, P. Zhang, 451
 and J. M. Dai, *Appl. Phys. Lett.* **101**, 241901 (2012). 452
³¹S. Michael and T. Ashutosh, *J. Appl. Phys.* **101**, 124912 (2007). 453
³²H. Mizoguchi, H. W. En, and P. M. Woodward, *Inorg. Chem.* **43**, 1667 454
 (2004). 455
³³R. D. Shannon, *Acta Crystallogr. A* **32**, 751 (1976). 456
³⁴M. J. Han, K. Jiang, J. Z. Zhang, W. L. Yu, Y. W. Li, Z. G. Hu, and J. H. 457
 Chu, *J. Mater. Chem.* **22**, 18463 (2012). 458

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